

Regiospecific Introduction of Amino-alkene Functionality into 1,2,3-Triols, 1,3-Dihalogenopropan-2-ols, and 2,3-Dihalogenopropanols promoted by Fluoride Anion

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Regiospecific transformations of 1,2,3-triol, 1,3-dihalogenopropan-2-ol, and 2,3-dihalogenopropanol derivatives into oxazolidin-2-ones and/or oxazol-2(3*H*)-ones promoted by fluoride anion are described.

We report a group of new chemo- and regio-specific transformations which use the naked fluoride anion¹ as a chemo-selective base to initiate the conversion of 1,2,3-triol derivatives into oxazolidinones and oxazolones² that are useful synthons and precursors for a variety of amino ketones, aldehydes, and alcohols.

The carbamate (**1a**)³ was treated with tetra-*n*-butylammonium fluoride (TBAF) and methanesulphonyl fluoride (MsF)⁴ in tetrahydrofuran (THF) at 50 °C to give the oxazolidinone (**3a**) exclusively. † This compound was readily converted into the aminomethyl ketone (**5a**) on treatment with sodium methoxide (1.1 equiv.) in methanol at room temperature. Results for other reactions of compounds (**1**) with TBAF–MsF are summarized in Table 1.

We next found that the dihalogeno compounds (**2**) underwent a similar type of reaction with TBAF to give methyleneoxazolidinone (**4**) and/or methyloxazolone (**7**) in good yields.

When R = alkyl in (**2**), the reaction gave the exocyclic methylene product (**4**) exclusively or predominantly, whereas aryl carbamates (R = Ar) gave the exocyclic methylene compound (**4**) or the ring-unsaturated isomer (**7**) selectively,

depending on temperature. At room temperature (**4**) was formed exclusively, whereas (**7**) was the sole product at 50 °C. Subsequent methanolysis afforded aminomethyl ketones in good yields.

An interesting selectivity was also observed in the case of the 2,3-dihalogenopropyl phenylcarbamate (**8**). Treatment of (**8**; X = Cl) with TBAF at room temperature gave (**10**) exclusively, whereas at 50 °C the exocyclic methyleneoxazolidinone (**9**) was obtained. In contrast to the chloride (**8**; X = Cl), the bromo analogue (**8**; X = Br) gave the exocyclic isomer (**9**) selectively regardless of temperature. This selectivity provides a simple alternative preparation of (**9**) and (**10**) to that previously reported.⁵ Compound (**10**) was also readily converted into the amino-aldehyde (**11**) in 62% yield upon treatment with sodium methoxide in methanol. Reactions of 1,3- or 2,3-dihalogeno derivatives are listed in Tables 2 and 3.

PhNCO₂CH(CH₂Br)CH₂CH₂Br and PhSO₂CH₂CO₂CH(CH₂Cl)₂ also participate in this type of reaction to give the cyclized products (**12**) (82%) and (**13**) (65%), respectively.

The phase transfer catalyst–KF system⁶ offers further interesting selectivity; the reaction ceased at the cyclization stage, and no unsaturated product was formed under these

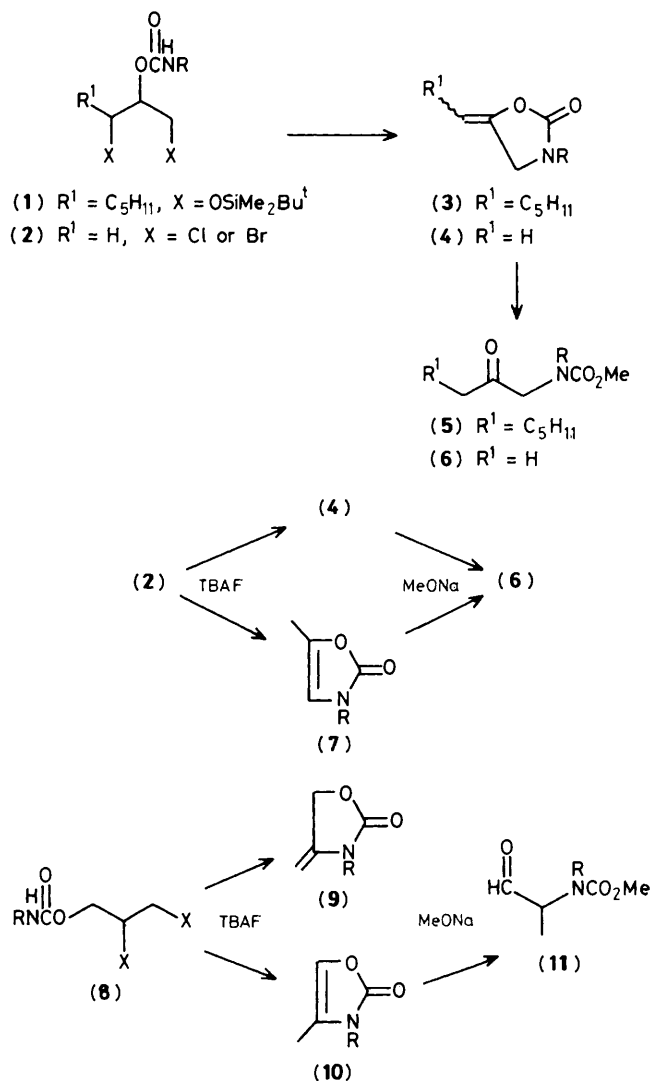
† A typical procedure is as follows: to a solution of TBAF·3H₂O, (1.75 mmol; Aldrich, used without further purification) and molecular sieves 4A (1.0 g) in THF (2 ml) was added a mixture of (**1a**) (0.30 mmol) and MsF (1.14 mmol) in THF (2 ml) at room temperature, and the mixture was stirred at that temperature for 2 h and then at 50 °C for 5.5 h. The mixture was filtered, dried (MgSO₄), and evaporated to give a crude oil (92 mg). Purification on preparative t.l.c. (eluant: AcOEt–*n*-hexane, 1 : 2) gave (**3a**) (49 mg, 67%) as a mixture of *E*- and *Z*-isomers. These isomers isomerized to 5-hexyl-3-phenyl-2(3*H*)-oxazolone on treatment with activated silica gel in *n*-hexane at room temperature.

Table 1. Reactions of compounds (**1**).^c

	R	(2), % yield ^a (<i>E</i> : <i>Z</i>) ^b	(3), % yield ^a
a;	Ph	67 (75 : 25)	76
b;	<i>c</i> -C ₆ H ₁₁	64 (63 : 37)	80
c;	Me	50 (68 : 32)	85

^a Isolated yield. All compounds gave satisfactory spectral data.

^b Determined by n.m.r. measurement of the alkene protons. ^c An 82 : 18 mixture of *anti* and *syn* isomers of (**1**). See ref. 3.

**Table 2.** Reactions of compounds (2).^a

a;	R	X	% Yield ^b		
			(4)	(7)	(6)
a;	Ph	Cl ^c	76	0	80
		Cl	0	72	88
		Br	0	79	—
		F	0	9	—
b;	4-ClC ₆ H ₄	Cl	0	65	—
		Br	0	71	—
c;	1-Naphthyl	Cl	0	75	—
d;	c-C ₆ H ₁₁	Cl ^c	90	0	77
		Cl	78	12	—
e;	Me	Cl	72	0	80

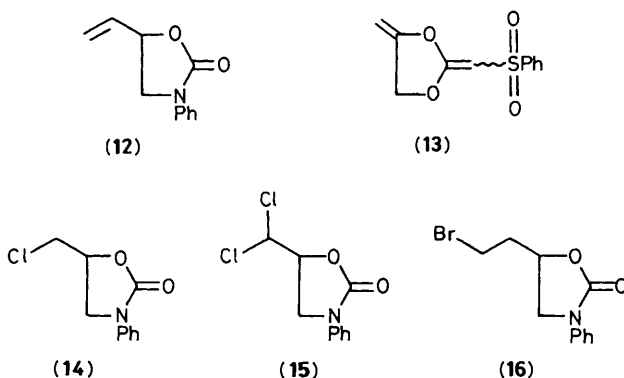
^a Reactions were carried out with (2): TBAF = 1.0:2.5–3.0 in THF at 50°C unless otherwise noted. The methanolysis was performed with 1.1 equiv. of MeONa in MeOH at 50°C unless otherwise noted. ^b Isolated yield. All compounds gave satisfactory spectral data. ^c At room temperature.

conditions: e.g. (2; R = Ph, X = Cl), PhNCO₂CH(CH₂Cl)-CHCl₂, and PhNCO₂CH(CH₂Br)CH₂CH₂Br gave (14) (83%), (15) (91%), and (16) (74%), respectively with 10 mol% of benzyltriethylammonium chloride and 4 equiv. of KF in toluene–water at 50–90°C.

Table 3. Reactions of compounds (8).

R	X	Temp./°C	% Yield ^b	
			(9)	(10)
Ph	Cl	R.t.	0	51
	Cl	50	69	0
	Br	R.t.	80	0
	Br	50	68	0
c-C ₆ H ₁₁	Cl	R.t.	0	17

^a Reactions were carried out with (8): TBAF = 1.0:2.5–3.0 in THF at room temp. or 50°C. ^b Isolated yield. All compounds gave satisfactory spectral data.



Thus, TBAF is an excellent chemoselective base for the introduction of amino-alkene functionality into 1,2,3-triol, 1,3-dihalogenopropan-2-ol, and 2,3-dihalogenopropanol derivatives, offering a versatile synthesis of oxazolidinones and oxazolones. Use of KF in a phase transfer catalyst system furnishes a selective approach to halogeno-oxazolidinones.

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References

- E. J. Corey and A. Venkateswarlu, *J. Am. Chem. Soc.*, 1972, **94**, 6190; W. Prescott, *Chem. Ind. (London)*, 1978, 56; M. R. C. Gerstenberg and A. Haas, *Angew. Chem., Int. Ed. Engl.*, 1981, **20**, 647; S. Rozer and R. Filler, *Tetrahedron*, 1985, **41**, 1111; I. Kuwajima and E. Nakamura, *Acc. Chem. Res.*, 1985, **18**, 181; J. H. Clark, *Chem. Rev.*, 1980, **80**, 429, and references therein; G. Cardillo, M. Orena, G. Porzi, S. Sandri, and C. Tomasini, *J. Org. Chem.*, 1984, **49**, 701.
- For the chemistry of these heterocycles, see: G. V. Boyd, in 'Comprehensive Heterocyclic Chemistry,' eds. A. R. Katritzky and C. W. Rees, Pergamon, Oxford, 1984; vol. 6, pp. 177–233, and references therein.
- Prepared *via* osmium tetroxide oxidation of 3-t-butylidimethylsiloxyoct-1-ene followed by selective t-butylidimethylsilylation and carbamoylation. The *anti:syn* ratio of the products was determined to be 82:18 by transforming the desilylated diols into the acetonide; cf. J. K. Cha, W. J. Christ, and Y. Kishi, *Tetrahedron Lett.*, 1983, **24**, 3943; W. J. Christ, J. K. Cha, and Y. Kishi, *ibid.*, 1983, **24**, 3947.
- M. Shimizu, Y. Nakahara, and H. Yoshioka, *Tetrahedron Lett.*, 1985, **26**, 4207; 1987, **28**, in the press; *J. Chem. Soc., Chem. Commun.*, 1986, 867; M. Shimizu, E. Tanaka, and H. Yoshioka, *ibid.*, 1987, 136.
- K. Sisido, K. Hukuoka, M. Tuda, and H. Nozaki, *J. Org. Chem.*, 1962, **27**, 2663; N. Shachat and J. J. Bagnell, Jr., *ibid.*, 1963, **28**, 991; P. J. Stoffel and W. D. Dixon, *ibid.*, 1964, **29**, 978.
- D. Landini, F. Montanari, and F. Rolla, *Synthesis*, 1974, 428; T. Kitazume and N. Ishikawa, *Chem. Lett.*, 1978, 283; M. Tordeux, S. R. Flecher, and I. T. Kay, *J. Chem. Soc., Chem. Commun.*, 1978, 903; C. Wakselman, *Synth. Commun.*, 1982, **12**, 513.